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Worksheet 5 Answer Key



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Nomenclature Worksheet 5: Ion~c  
Compounds Summary Name the following  
compounds: Give the formula for each  
compound: 1.  $\text{CaF}_2$  2.  $\text{Na}_2\text{O}$  3.  $\text{BaS}$  4.  
 $\text{CuSO}_4$  5,  $\text{Fe}_2\text{O}_3$  6,  $\text{HgCl}_2$  7.  $\text{AgNO}_3$  8.  
 $\text{MgCO}_3$  9.  $\text{K}_2\text{C}_2\text{H}_3\text{O}_2$  10.  $\text{K}_2\text{Cr}_2\text{O}_7$  11  
 $\text{Al}(\text{OH})_3$  12.  $\text{PbBr}_2$  13.  $\text{ZnSO}_3$   
14.  $\text{NaHCO}_3$  15.  $\text{NH}_4\text{Cl}$  16.  $\text{Li}_3\text{PO}_4$

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17. SnC<sub>2</sub> 18. Al(NO<sub>2</sub>)<sub>3</sub> 19. Rb<sub>2</sub>CrO<sub>4</sub> 20.  
KMnO<sub>4</sub> 21. CuCl 22. FeSO<sub>4</sub> 23. sodium  
fluoride 24. potassium sulfide

Nomenclature Worksheet 5: Ion-  
Compounds Summary

Naming Part Worksheet - 5 Determine the  
best naming part to complete each

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sentence. ID: 1018318 Language: English  
School subject: grammar Grade/level: 1  
Age: 5-7 ... Check my answers: Email my  
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style: Font: Size: px. Font ...

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Worksheet 5.1 Writing and Naming Ionic  
Compounds with Polyatomic Ions and

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Transition Metals. Section A Write the name of the ionic compounds containing polyatomic ions. (BE CAREFUL TRANSITION METALS MAY HAVE ROMAN NUMERALS and NICKNAMES)

1.  $\text{Pb}_3(\text{PO}_4)_2$  lead(II) phosphate
2.  $\text{Cu}_2\text{CO}_3$  copper (I) carbonate
- 3.

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Answer Key Worksheet 5.1 Naming and  
Writing Ionic ...

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Covalent (Molecular) Compounds Part A:  
Name the following covalent. compounds.



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1. CO Carbon monoxide .
2. CO<sub>2</sub> Carbon dioxide.
3. N<sub>2</sub>O<sub>3</sub> Dinitrogen trioxide.
4. N<sub>2</sub> omit.
5. NP Nitrogen monophosphide.
6. SCl<sub>2</sub> Sulfur dichloride.
- 7.

Nomenclature Worksheet 2 Answer Key  
Covalent Nomenclature Worksheet

□ Answer Key. Write the correct name for:

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1)  $\text{As}_4\text{O}_{10}$ . tetrasenic decoxide. 2)  $\text{BrO}_3$ .  
bromine trioxide. 3)  $\text{BN}$ . boron  
mononitride. 4)  $\text{N}_2\text{O}_3$ . dinitrogen trioxide.  
5)  $\text{NI}_3$ . nitrogen triiodide. 6)  $\text{SF}_6$ . sulfur  
hexafluoride. ... Covalent Nomenclature  
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oxidation number, valence electrons.

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