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Measurement of

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Measuring

Mass. a measure of how much matter is in an object. Volume. the amount of space that matter occupies.

Density. the measurement of how much mass of a substance is contained in a given volume.

International System of Units (SI) the system of units used by scientists to measure the

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properties of matter.

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Measure the lines to the nearest millimeter and to the nearest tenth of a centimeter (2.3 cm). _____

71 mm 7.1 cm.

Within 1mm is acceptable answer.

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Science!
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false 2. true 3. false 4.

true 5. true 6. true 7.

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Class: _____ 1.

Count the number of significant digits in the following numbers: a.

863.00 b. 307 c.

0.000591 d. 20.00 e.

3000 2. Perform the following calculations and report your answer in correct number of significant figures: a.

$87.339 + 23.1 =$ b.

$42.67 - 11.27 =$ c. $12 \times$

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12 = d. 5.4 / 2.02 = e.

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Mole Section 101

Measuring Matter 1

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moles 11 Chapter 10

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Answer Key

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Unit 1: Measuring

Matter and Energy

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matter? [TYPE

ANSWER HERE] 2.

Define density in a

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complete sentence.

[TYPE ANSWER
HERE] 3. Is density a

physical or chemical
property of matter?

Explain what this
means. [TYPE

ANSWER HERE] 4.

What is the

mathematical definition
of density? [TYPE

ANSWER HERE] 5.

What are the units used
to define mass?

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The percent composition of a compound is the

percent by mass of each of the elements in a compound. 2. Divide the mass of each element in the sample by the mass of the sample. Then multiply each quotient by 100. 3.c.

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wine become the Body

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The chemical

compound CH_2O is

organic matter; CO_2 is

carbon dioxide gas; O_2

is diatomic oxygen; and

H_2O is water. The

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atomic weights of the elements C, H, and O are, respectively, 12, 1, and 16. A. What are the gram molecular weights of the compounds of CH_2O , CO_2 , O_2 ...

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